



Libyan International Medical University

Faculty of Pharmacy

Potentiometry

Amena Hammoda Takwa Esam Faiza Shahut

Objectives

- Define potentiometry
- Explain components of potentiometry
- List types of potentiometry
- Identify uses of potentiometry

Introduction



In this presentation we will define potentiometry which is a technique similar to direct titration, then the components of potentiometry. Moreover, list the four types Potentiometry and their uses.

Define potentiometry

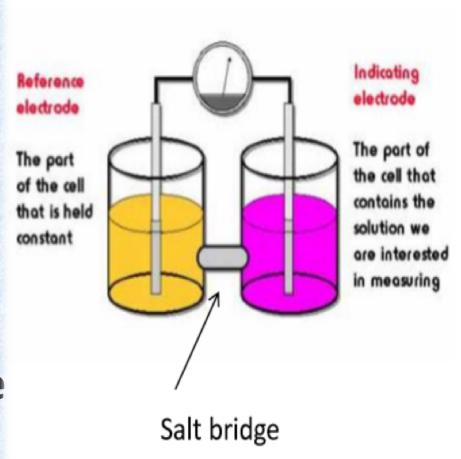


Potentiometry is one of the methods of electroanalytical chemistry. It is usually employed to find the concentration of a solute

in solution.

Explain components of a Potentiometric Cell

- > Analyte
- **≻**Salt bridge
- > Indicator electrode
- > Reference electrode



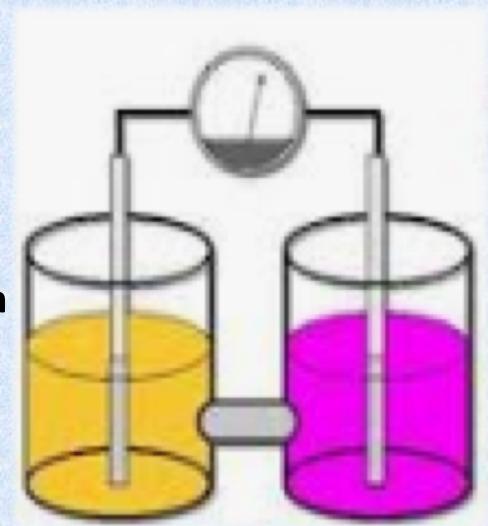
List types of potentiometry

1-Redox titration

2-Acid-base titration

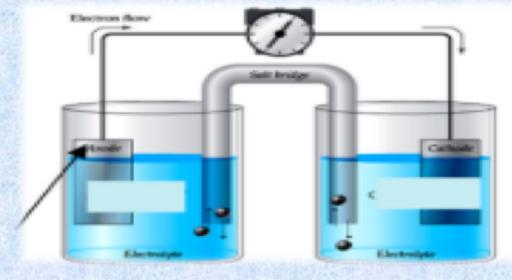
3-Complexmetric titration

4-Precipitation titration



Identify uses of potentiometry

 Potentiometric titration is used for measuring the change in the electrical potential when a neutralizing agent (titrant) is added to a chemical solution.



Summary



 We talked about Potentiometry which is one of the methods of electroanalytical chemistry and then mentioned its components. In addition, we listed the types of potentiometry and their uses.

Reference

Snustad, D.P. & Simmons, M.J. (n.d.) *Analytical Chemistry 7e by Gary D. Christian*.

Anon (n.d.) potentiometry. [Online] Available from: https://brainly.in/question/1619257.

